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Low-Temperature Tunneling in the Spin-State **Relaxation Dynamics of Iron(III) Spin-Crossover** Compounds

Sabine Schenker and Andreas Hauser*

Institut für anorganische und physikalische Chemie Universität Bern, Freiestrasse 3 3000 Bern 9, Switzerland

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Some 13 years ago Buhks et al.¹ suggested that the spin conversion in spin-crossover compounds should be treated as a nonadiabatic process within the theory of radiationless multiphonon relaxation. They predicted a thermally activated highspin \rightarrow low-spin (HS \rightarrow LS) relaxation at elevated temperatures and a nonvanishing tunneling rate at cryogenic temperatures. Although Buhks et al. refer to spin-crossover complexes in solution, the theory is equally valid for the solid state,² with the low-frequency lattice vibrations playing the role of the heat bath instead of solvent librations.

For the HS(${}^{5}T_{2}$) \rightarrow LS(${}^{1}A_{1}$) relaxation in Fe(II) compounds, Xie and Hendrickson subsequently found the predictions to be true,³ and Hauser et al. demonstrated the validity of the inverse energy gap law in the strong vibronic coupling limit.⁴ This served to explain the extremely long-lived light-induced HS states with low-temperature lifetimes of 102-106 s in these compounds.4.5

Lawthers and McGarvey showed a perturbation of the LS- $(^{2}T_{2}) \Leftrightarrow HS(^{6}A_{1})$ equilibrium in Fe(III) compounds to be possible with pulsed laser excitation, too.⁶ Despite this and the fact that at ambient temperatures the HS \rightarrow LS relaxation rate constants for Fe(III) spin-crossover compounds are typically only an order of magnitude larger than for Fe(II) systems, i.e., 10^{8±1} s⁻¹ versus $10^{7\pm1}$ s⁻¹,⁷ no reports of long-lived metastable states at low temperatures for the former have appeared in the literature so far.

The Fe(III) spin-crossover compound $[Fe(acpa)_2]PF_6(Hacpa$ = N-(1-acetyl-2-propylidene)-N-(2-pyridylmethyl)amine)⁸ is ideally suited for a photophysical investigation of the HS \rightarrow LS relaxation down to low temperatures. Figure 1 shows the temperature dependence of the absorption spectrum between 150 and 300 K in propriononitrile/butyronitrile (4:5). The LMCT band at 14 600 cm⁻¹, characteristic of the LS species, disappears at elevated temperatures and is replaced by the characteristic HS band at 17 600 cm⁻¹. The transition temperature $T_{1/2}$ of \sim 200 K is slightly higher than the \sim 180 K in the solid state.⁹ Pulsed laser excitation within the LS band ($\nu = 14400 \text{ cm}^{-1}, 3$ mJ/pulse, 7-ns pulse width) results in a transient bleaching at 14 800 cm⁻¹ and a corresponding increase in absorption at 18 450 cm⁻¹, indicating that we are indeed looking at a light-induced population of the HS state followed by HS \rightarrow LS relaxation. The inset in Figure 2 shows the relaxation curves at 60 K for [Fe- $(acpa)_2$]PF₆ dispersed in KBr. They are slightly nonexponential,

Xie, C.-L.; Hendrickson, D. N. J. Am. Chem. Soc. 1987, 109, 6981.
Hauser, A.; Vef, A.; Adler, P. J. Chem. Phys. 1991, 95, 8710.
(a) Decurtins, S.; Gütlich, P.; Köhler, C. P.; Spiering, H.; Hauser, A.

- Chem. Phys. Lett. 1984, 105, 1. (b) Decurtins, S.; Gütlich, P.; Hasselbach, K. M.; Spiering, H; Hauser, A. Inorg. Chem. 1985, 24, 2174. (c) Hauser, A.;
- Adler, J.; Gütlich P. Chem. Phys. Lett. 1988, 152, 468. (6) Lawthers, I.; McGarvey, J. J. J. Am. Chem. Soc. 1984, 106, 4280.

(7) Beatile, J. K. Adv. Inorg. Chem. 1988, 32, 1.
(8) Maeda, Y.; Oshio, H.; Takashima, Y.; Mikuriya, M.; Hidaka, M. Inorg. Chem. 1986, 25, 2958.

(9) Oshio, H.; Toriumi, K.; Maeda, Y., Takashima, Y. Inorg. Chem. 1991, 30. 4252.

150 K 160 170 180 200 210 220 230 240 250 260 270 - 293 600 500 [II / mol cm] 400 300 200 100 15000 20000 25000 30000 [cm`1]

Figure 1. Absorption spectrum of [Fe(acpa)₂]PF₆ in propriononitrile/ butyronitrile (4:5) between 150 and 300 K.



Figure 2. $\ln[k_{\rm HL}(T)]$ versus 1/T for the HS \rightarrow LS relaxation in [Fe- $(acpa)_2]PF_6$ in KBr. Effective activation energy at high temperatures E_a = 730(40) cm⁻¹. Inset: transient absorption of $[Fe(acpa)_2]PF_6$ in KBr at 60 K monitored at (a) 18 450 and (b) 14 800 cm^{-1} after pulsed laser excitation at 14 400 cm⁻¹.

probably due to small laser-induced temperature gradients and inhomogeneities in the matrix rather than to the physics of the relaxation process as such. Mean relaxation rate constants are plotted as $\ln[k_{HL}(T)]$ versus 1/T in Figure 2. Evidently, the same type of behavior as for the HS \rightarrow LS relaxation in Fe(II) spincrossover systems is observed, namely, more or less temperature independent tunneling below 100 K, and a thermally activated process at elevated temperatures. For [Fe(acpa)₂]PF₆ in KBr, the low-temperature tunneling rate constant $k_{\rm HL}(T \rightarrow 0)$ is $\sim 10^2$ s^{-1} , the activation energy for the thermally activated process E_a is \sim 730(40) cm⁻¹. Above 200 K the relaxation rate constant is $>5 \times 10^6$ s⁻¹, in accordance with the observed line broadening of Mössbauer spectra.9

How does the value of 10² s⁻¹ for the low-temperature tunneling rate constants compare to the 10⁻⁶-10⁻² s⁻¹ typically observed for Fe(II) spin-crossover compounds?^{4,5} According to the equation²

$$k_{\rm HL}(T \rightarrow 0) = \frac{2\pi}{\hbar^2 \omega} g_{\rm f} \beta_{\rm HL}^2 |\langle \chi_{\rm p} | \chi_{\rm o} \rangle|^2 = \frac{2\pi}{\hbar^2 \omega} g_{\rm f} \beta_{\rm HL} \frac{2S^p \, {\rm e}^{-S}}{p!} \quad (1)$$

where β_{HL} is the electronic matrix element given by spin-orbit coupling, g_f the electronic degeneracy of the final state, and $|\langle \chi_p | \chi_o \rangle|^2$ a Franck-Condon factor, $k_{\rm HL}(T \rightarrow 0)$ depends critically upon the relative vertical and horizontal displacements of the potential wells of the initial and the final state. The former is conveniently expressed by the reduced energy gap p = $\Delta E_{\rm HL}^0/\hbar\omega$, the latter by the Huang-Rhys factor $S = 1/2f\Delta Q^2/2$ $\hbar\omega$, where ΔQ is the difference in equilibrium position in the

⁽¹⁾ Buhks, E.; Navon, G.; Bixon, M.; Jortner, J. J. Am. Chem. Soc. 1980, 102, 2918.

⁽²⁾ Donnelly, C. J.; Imbush, G. F. NATO ASI Ser. B 1991, 245, 175.

accepting mode, f and $\hbar\omega$ are the corresponding force constant and vibrational frequency, respectively, and $\Delta E_{\rm HL}^0$ is the zeropoint energy difference.

For Fe(II) compounds with an [FeN₆] core, a value of S of ~45 has been estimated from the difference in the metal-ligand bond length between the HS and the LS state $\Delta r_{\rm HL}$ of ~0.16– 0.21 Å,¹⁰ an average force constant f of ~2 × 10⁵ dyn/cm, and a vibrational frequency $\hbar \omega$ for the totally symmetric mode of ~250 cm^{-1,4} For Fe(III) systems with an [FeN₄O₂] core, this is not as straightforward. However, with an average value of $\Delta r_{\rm HL}$ of ~0.12 Å,⁹ a 50% larger vibrational frequency, and a correspondingly larger force constant, a value for S of ~25 is a reasonable estimate. For spin-crossover compounds with $T_{1/2}$ between 100 and 200 K, the reduced energy gap p is ~1, both for Fe(II) and Fe(III) systems. Thus with the electronic matrix element for Fe(II) being ~150 cm⁻¹ and for Fe(III) ~40 cm⁻¹,¹ the relation

$$\frac{k_{\rm HL}^{\rm II}(T \rightarrow 0)}{k_{\rm HI}^{\rm II}(T \rightarrow 0)} \approx 10^6 - 10^7 \tag{2}$$

holds, which is in good agreement with observation. Of course, this is only a very crude estimate, but it shows that basically due to the smaller value of $\Delta r_{\rm HL}$ for Fe(III) the much larger low-temperature tunneling rate constants can be rationalized at least semiquantitatively. At elevated temperatures the ratio of $k(T)_{\rm HL}^{\rm HI}$ to $k(T)_{\rm HL}^{\rm HI}$ becomes smaller, and for a meaningful estimate the actual vibrational frequencies and the exact bond length and zero-point energy differences of the complexes in hand would have to be explicitly considered.

Further work, both experimental and theoretical, is needed to understand the HS \rightarrow LS relaxation in spin-crossover compounds, and thus intersystem crossing dynamics in the strong vibronic coupling limit, at the required level.

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^{(10) (}a) Hoselton, M. A.; Wilson, L. J.; Drago, R. S. J. Am. Chem. Soc. 1975, 97, 1722. (b) Katz, B. A.; Strouse, C. E. J. Am. Chem. Soc. 1979, 101, 6214. (c) Mikami, M.; Saito, Y. Acta Crystallogr. 1982, B38, 452.